

REVIEW

Arylamine N-acetyltransferases: a structural perspective

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Arylamine N-acetyltransferase (NAT) plays an important role in metabolism and detoxification of many compounds including drugs and environmental carcinogens through chemical modification of the amine group with an acetyl group. Recent studies have suggested that NATs are also involved in cancer cell growth and inhibition of the enzymes may be a potential target for cancer chemotherapy. Three-dimensional (3D) structures are available for NATs from both prokaryotes and eukaryotes. These structures provide valuable insights into the acetylation mechanism, features of the active site and the structural determinants that govern substrate/inhibitor-binding specificity. Such insights allow a more precise understanding of the structure–activity relationships for NAT substrates and inhibitors. Furthermore, the structural elucidation of NATs has generated powerful tools in the design of small molecule inhibitors that should alleviate cancer, based on the important role of the enzyme in cancer biology.

Abbreviations

3D, three-dimensional; 5-AS, 5-aminosalicylic acid; ANS, anisidine; CoA, Coenzyme A; HDZ, hydrazines; NAT, Arylamine N-acetyltransferase; PABA, p-aminobenzoic acid; PAS, p-aminosalicylic acid; SMZ, sulfamethazine; TZD, thiazolidinedione

Introduction

Arylamine N-acetyltransferases (NATs; EC 2.3.1.5) catalyse the acetylation reaction (generally classified as a Phase II process) by which an acetyl group from the cofactor acetyl coenzyme A (acetyl CoA) is transferred to the substrates, including drugs and carcinogens. NATs are present in most of living organisms (Butcher et al., 2002; Vagena et al., 2008) and NAT proteins have a characteristic motif - (Pro/Ala)-Phe-Glu-Asn-Leu [(P/A)FENL] (Payton et al., 2001). Although its endogenous functions remain largely unclear, NAT is essential for synthesis of cell wall in mycobacteria (Sim et al., 2008a). Human NATs include two isoenzymes, namely, NAT1 and NAT2. NAT1 is found in many tissues, particularly in the colon, whereas NAT2 is found mostly in the liver and intestine and is predominantly responsible for drug metabolism (Hickman et al., 1998; Sim et al., 2000). It acetylates many therapeutic agents such as isoniazid, dapsone and the sulphonamides.

In recent years, acetylation has been intensively studied due largely to its role in the carcinogenic activation of aromatic amines (Hein *et al.*, 1993). These acetylated derivatives of arylamines and N-hydroxylated arylamines can undergo further oxidation or other reactions to form highly reactive cytotoxic and carcinogenic species (Hein *et al.*, 1993; Hanna, 1994). Moreover, there is increasing evidence that NAT has an important role in folate catabolism and cancer cell biology (particularly for breast cancer) (Adam *et al.*, 2003; Butcher *et al.*, 2007; Sim *et al.*, 2008b; Butcher and Minchin, 2012). In fact, human NAT1 has emerged as a new diagnostic marker or drug target for breast cancer (Laurieri *et al.*, 2010; Russell *et al.*, 2009; Tiang *et al.*, 2010).

The first X-ray crystal structure for a NAT enzyme (from *Salmonella typhimurium*) was reported by Sinclair *et al.* (2000). To date, there are 11 crystal structures for NATs from human and five bacterial species (Table 1). This structural bank has provided valuable insights into the acetylation mechanisms and active site features. Importantly, it helps to explain why



 Table 1

 List of available crystal structures for NATs (the structures can be found at http://www.pdb.org)

NAT	PDB entries	In complex with	Reference
S. typhimurium NAT	1E2T		Sinclair et al., 2000
M. smegmatis NAT	1GX3		Sandy <i>et al.</i> , 2002
M. smegmatis NAT	1W6F	Isoniazid	Sandy <i>et al.</i> , 2005b
M. smegmatis NAT (C70Q)	1W5R		Sandy et al., 2005a
Pseudomonas aeruginosa NAT	1W4T		Westwood et al., 2005
Mesorhizobium loti NAT1	2BSZ		Holton <i>et al.,</i> 2005
M. marinum NAT	2VFB		Fullam <i>et al.</i> , 2008
M. marinum NAT	2VFC	Coenzyme A	Fullam <i>et al.</i> , 2008
Human NAT1 (F125S)	2IJA	Iodoacetamide	Wu <i>et al.,</i> 2007
Human NAT1	2PQT	2-Bromoacetanilide	Wu <i>et al.,</i> 2007
Human NAT2	2PFR	Coenzyme A	Wu et al., 2007

NATs selectively catalyse aromatic amines and exhibit overlapping yet distinct substrate specificity. Both human NAT1 and NAT2 are highly polymorphic enzymes. Many of the NAT polymorphisms translate to mutations in particular protein residues. The crystal structures have suggested individual function of the residues in the protein that has greatly improved the understanding of the phenotypes associated with NAT polymorphisms. The crystal structures represent not only a tool for predicting substrate selectivity but also a powerful tool in rational design of selective inhibitors of NATs for therapeutic purposes.

NAT Structure

The secondary and tertiary structures of the NATs are highly conserved among the enzymes from prokaryotes and eukaryotes (Westwood et al., 2006; Sim et al., 2008a). The NAT fold is usually described in terms of three domains (Sinclair et al., 2000; Wu et al., 2007). For example, in human NAT, the N-terminal domain [or the helical bundle in Sinclair et al., (2000)] consists of five helices ($\alpha 1-\alpha 5$) and one short β-strand between helices α2 and α3 (Figure 1). The second domain [or the β -barrel in Sinclair *et al.*, (2000)] consists of 10 β -strands (β 2– β 11) and two short helices α6 and α7 (Figure 1). The two domains connect through the α -helical interdomain (helices $\alpha 8$ – $\alpha 10$) to the third domain [or the α/β lid in Sinclair et al., (2000)], which has four anti-parallel β -strands (β 12– $\beta15$) and helix $\alpha11$ (Figure 1). The helix $\alpha11$ precedes a stretch of residues that lead across the protein molecule's surface into a buried C-terminus (Figure 1).

Although a highly similar structural fold is shared by bacterial and human NATs, structural differences are evident. First, human NATs contain a 17-residue insertion (i.e. residues 167–183) between $\beta 9$ and $\beta 11$, which is absent in the structures of prokaryotic NATs (Wu *et al.*, 2007) (Figure 2). The insertion includes the secondary structure elements $\beta 10$ and $\alpha 7$. Although the function of this insertion in human NATs is

not fully understood, the insertion does contribute to the stability of the protein by interacting with surrounding residues (Walraven *et al.*, 2007). Second, human NATs show a significant structural divergence in the arrangement of their C-terminal residues compared with prokaryotic NATs (Figure 2). In human NAT, the C-terminal residues form a full coil and the terminus extends deep within the folded protein into close proximity to the buried catalytic triad and is capable of interacting with CoA (Wu *et al.*, 2007) (Figure 2). By contrast, the C-terminus in prokaryotic NATs is in a helical configuration, and the helix is positioned away from the protein core (Figure 2).

Active sites for cofactor and substrate binding

The crystal structures, in complex with CoA, for Mycobacterium marinum NAT and human NAT2 reveal that CoA binds differently to these two enzymes (Wu et al., 2007; Fullam et al., 2008). In human NAT, the CoA-binding site is located in a deep cleft formed between the interdomain (connecting domain II to domain III) helices $\alpha 8-10$ and the β -barrel (domain II) (Wu et al., 2007) (Figure 3A). CoA is bound to the active site in a bent conformation, positioning the sulfhydryl group of CoA towards the protein core and the 3'-phospho-ADP group on the surface of the protein (Figure 3A). The β -mercaptoethylamine and pantothenate moieties are located deep inside the cleft, placing the cofactor's sulfhydryl group close to that of the catalytic cysteine (C68), with a sulfur-sulfur distance of ~3 Å (Wu et al., 2007). Binding of CoA to the active site is mainly mediated by hydrophobic and hydrogen-bond interactions (Figure 3B). The pantotheine arm of CoA makes an extensive set of van der Waals contacts with the non-polar residues F37, F93, L209, F217 and L288 (Figure 3B and C). The adenine ring also makes contacts with the hydrophobic residues P97 and V98. In addition, the carbonyl group of the pantothenate moiety forms a hydrogen bond with the nitrogen atom of S216. The pyrophosphate group of CoA makes a series of hydrogen bonds with the residues T103, G104, Y208 and T214. The N6 of the adenine

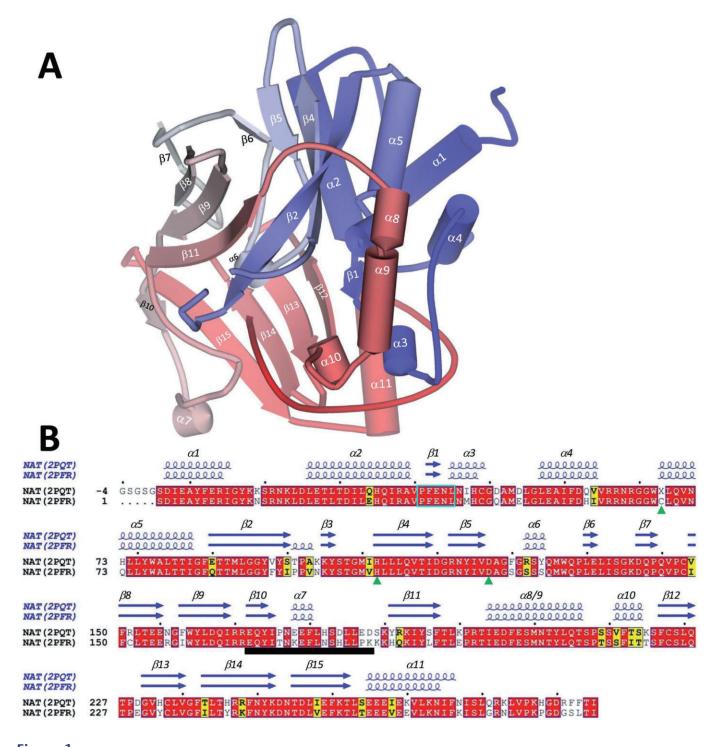


Figure 1

The tertiary and secondary structures of NAT. Panel A: The tertiary structure of NAT (using human NAT1 (PDB code: 2PQT) as an example), depicting the overall fold of a NAT structure. Panel B: Structure-based sequence alignment of human NATs (NAT1 and NAT2). The secondary structure elements are shown above the alignment. Conserved residues are highlighted in colour. The Protein Data Bank (PDB) codes for the structures are shown in parentheses. The triangles indicate the catalytic residues (please see text for details). This figure was produced using ENDscript (Gouet and Courcelle, 2002).

ring of CoA forms a hydrogen bond with the side chain of S287 (Wu et al., 2007) (Figure 3B).

Unexpectedly, a different CoA-binding site is present in the prokaryotic NAT (Fullam *et al.*, 2008). Here, CoA binds to the deep cleft formed by domains II and III of the enzyme (Figure 4A). The binding site is located ~30 Å from that in human NAT (Figure 4). Recognition of the pantotheinate arm of CoA is mainly through hydrophobic interactions with



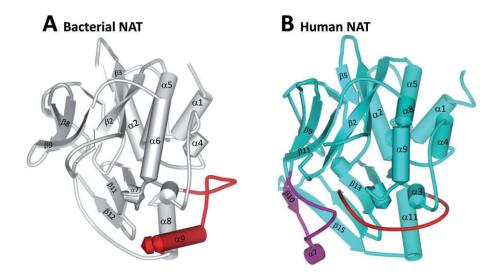


Figure 2
Structural comparison of bacterial NAT (panel A) and human NAT (panel B). The structures of NAT from Salmonella typhimurium (PDB code: 1E2T) and human NAT1 (PDB code: 2PQT) are shown. The insertion in the human NAT is indicated in purple. The structures of the C-terminal residues in both enzymes are shown in red.

residues F38, Y69, Y71, H110, F130 and F204. The phosphate groups of CoA are hydrogen-bonded to residues such as W97 and K236. The adenine moiety of CoA interacts with the active site residues via a mixture of hydrophobic and polar contacts. Specifically, it forms hydrophobic, hydrogen-bond and ring-stacking interactions with residues V169, E152, H229 respectively (Fullam *et al.*, 2008). The observation that human and prokaryotic NATs bind CoA in markedly different ways led Fullam *et al.* (2008) to hypothesize that the cofactor binding site may be a promising novel target for selectively inhibiting pathogenic prokaryotic NAT enzymes.

The acetyl acceptor (substrate) binding site overlaps to a great extent with the CoA-binding site and this finding is consistent with the fact that the cofactor and the substrate bind to the enzymes in a sequential manner, a feature of the Ping Pong kinetic mechanism (see discussion below). Similar to the pantotheine arm of CoA, the entire substrate molecule binds to the deep position of the cleft formed between the helical interdomain and domain II (β -barrel) (Wu *et al.*, 2007). The substrate-binding sites deduced from either prokaryotic or eukaryotic NATs consist mainly of hydrophobic residues (Figure 5A). This feature agrees with the fact that NATs have a preference for hydrophobic substrates.

Catalytic mechanism

There is strong evidence that the acetylation reaction mediated by NATs proceeds via the double displacement ('Ping Pong Bi Bi') mechanism (Andres *et al.*, 1983; 1988; Sinclair and Sim, 1997). There are two sequential steps to the reaction: firstly, the acetyl group is moved from acetyl CoA to form an acetylated enzyme intermediate, then the substrate is acetylated and CoA is released (Figure 6). Structural analysis reveals that the acetyl transfer in NATs is facilitated by a

Cys-His-Asp catalytic triad (Cys⁶⁸-His¹⁰⁷-Asp¹²² in human NAT), which is strictly conserved in all known NATs and in identical positions in all NAT structures (Sinclair et al., 2000). The triad renders a stable thiolated form of cysteine that initiates the reaction by nucleophilic attack on the carbonyl of the acetyl moiety of acetyl CoA (Figure 6) (Wang et al., 2004; Sandy et al., 2005a). Consistent with their critical role in catalysis, mutations of any of the triad residues abolishes the enzyme activity (Dupret and Grant, 1992; Watanabe et al., 1992; Wang et al., 2004). Interestingly, the latest NAT structure (from M. marinum) presents an extended binding interface between CoA and the protein (Fullam et al., 2008). The authors propose that the nucleosidephosphate moiety of CoA can remain associated with the protein, subsequent to acetylation of the catalytic cysteine, thereby protecting the acetylated enzyme intermediate from hydrolysis.

Wang et al. have proposed a model for the reaction chemistry in the NAT active site (Wang et al., 2004; 2005). In this model, a thiolate-imidazolium ion pair is formed between Cys⁶⁸ and His¹⁰⁷ with a pK_a of 5.2. Asp¹²² is involved in the ionic interaction with His 107 and is critical for optimal catalysis and structural integrity. Upon acetylation of the thiolate, half of the ion pair is lost with concomitant shift in the pKa of His¹⁰⁷ to 5.5. The process of nucleophilic attack on the thiol ester and deacetylation of the thiolate is dependent on the nucleophilic strength of the arylamine substrate. For weak nucleophiles with a pKa <5.5, such as p-aminobenzoic acid (PABA, Figure 7A), a deprotonation step by His¹⁰⁷ is required to make the substrates more nucleophilic for the attack on the thiol ester, which is followed by deacetylation of the thiolate. By contrast, those strong nucleophiles with a $pK_a \ge$ 5.5 such as anisidine (ANS, Figure 7A) directly attack on the thiol ester; deacetylation of the thiolate occurs via deprotonation of a tetrahedral intermediate (Wang et al., 2005) (Figure 6).

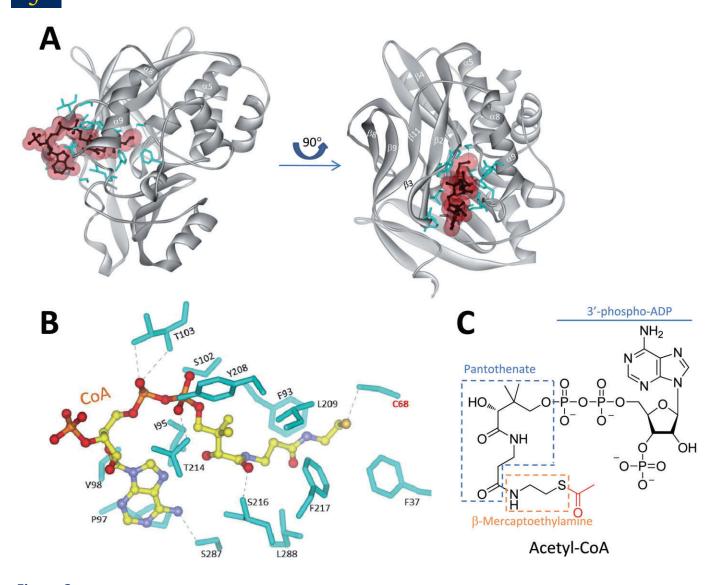


Figure 3 The CoA binding site in human NAT (PDB code: 2PFR), Panel A: A full view of CoA binding to human NAT, Panel B: Molecular interactions of CoA with the binding site residues. Panel C: Chemical structure of acetyl CoA, showing its three components.

Structure-activity relationships for **NAT** substrates

Human NAT1 and NAT2 exhibit an overlapping substrate specificity. Both enzymes display substrate preference for aromatic amines (Kawamura et al., 2005). This is largely explained by the hydrophobic feature of the active site and the presence of aromatic residues in the active site capable of forming ring-stacking interactions (Wu et al., 2007). However, distinct substrate specificity also exists between NAT1 and NAT2 (Kawamura et al., 2005).

Aminobenzyl compounds such as p-aminosalicylic acid (PAS; Figure 7A) were identified as binding preferentially to NAT1. In contrast, the sulfonamide class of compounds such as sulfamethazine (SMZ) bound selectively to NAT2 (Sim et al., 2008b). Wu et al. (2007) explored the molecular basis for this difference in substrate recognition between the two enzymes. The substrate binding pocket in NAT1 is smaller (162 Å³) than that of NAT2 (257 Å³) as a consequence of two key residue substitutions at positions 127 and 129, namely R127 and Y129 in NAT1, whereas in NAT2, serine residues occupy these positions (Figure 5B). The presence of two bulkier groups reduces the volume of the NAT1 pocket by ~40% compared with that of NAT2. Furthermore, V93 in NAT1 is replaced by F93 in NAT2; this substitution introduces a bump in the van der Waals surface of the pocket in NAT2, thereby significantly altering the shape of the binding pocket (Figure 5B). Molecular docking reveals that PAS and SMZ are closely fitted into the active sites of NAT1 and NAT2 respectively (Wu et al., 2007). The shape and size of the substrate molecules are well matched to those of the binding pockets, which helps to explain the distinct substrate selectivity of the enzymes.



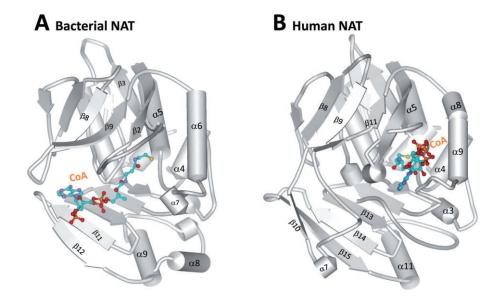


Figure 4 Comparison of the CoA-binding site in bacterial NAT from M. marinum (PDB code: 2VFC; panel A) with that in human NAT2 (PDB code: 2PFR; panel B).

A mutagenesis study showed that F125S substitution virtually abolished NAT1 affinity preference for PAS over SMZ and substrate selectivity of the mutant NAT1 (F125S) resembled that of wild-type NAT2 (Hein, 2002). Structural analysis and molecular docking indicate that F125 plays a key role in binding of PAS to NAT1 as the side chain of F125 forms π - π stacking interactions with the benzene ring of PAS (Wu et al., 2007). Therefore, F125S mutation would reduce the enzyme's affinity for the smaller PAS due to the lack of the π - π stacking interactions (as main forces) orienting the substrate for catalysis. On the other hand, an increase in the pocket space allows accommodation of the bulkier SMZ in the active site with more favourable energy. This is consistent with the fact that the mutant has an increased affinity for SMZ compared with the wild type (Hein, 2002).

Human NAT1 acetylates PABA and 4-aminobiphenyl but not o-toluidine (Grant et al., 1991; Hein et al., 1993; Fretland et al., 1997). The NMR structures of human NAT1 reported by Zhang et al. (2006) provide an explanation of such substrate selectivity. Based on the resolved structures, PABA binds favourably to the active site, consistent with the fact that it is a good NAT1 substrate (Figure 8A). However, binding of o-toluidine to the enzyme is difficult because its methyl group engages in steric clashes with F125 (Zhang et al., 2006). Also o-toluidine lacks functional groups such as the carboxylic acid group of PABA that could provide favourable binding contacts, for instance, hydrogen bonding (Zhang et al., 2006). The important role of F125 in determining the acetylation of o-toluidine is supported by the finding that a mutant human NAT2 in which F125 is replaced by a serine, exhibits 3- to 10-fold higher acetylation capacity for the substrate than NAT1 (Hein et al., 1993; Zhang et al., 2006). Based on this knowledge, Zhang et al. (2006) predicted that 4-amino-3methylbenzoic acid would be a worse NAT1 substrate than PABA due to the presence of the methyl group. Interestingly,

this prediction agreed well with the experimentally determined substrate activities. The authors also explained why 4-aminobiphenyl is acetylated by NAT1 by docking of the substrate with the enzyme (Zhang et al., 2006), as 4-aminobiphenyl was easily accommodated in the active site (Figure 8B). Binding of 4-aminobiphenyl was largely driven by the hydrophobic interactions between residues V93, K100, I106, F125, L209, S215, V216 and F217 and the aromatic ring of the substrate molecule (Figure 8B).

Human NAT2 displays a preference for lipophilic arylamines and the acetylation rate increases as the alkyl chain length of alkoxyanilines increases (Kawamura et al., 2005). On the contrary, human NAT1 displays a decrease in activity with increasing alkyl chain length (Kawamura et al., 2005). It has been proposed that this is due to the smaller overall size of human NAT1 active site compared with that of human NAT2 (Kawamura et al., 2005; Westwood et al., 2006). Substrate profiling demonstrates that none of the arylhydrazine compounds are substrates of human NAT1, whereas these compounds can be acetylated by human NAT2 (Kawamura et al., 2005). However, the reason why human NATs show this difference in substrate selection is unknown.

Brooke et al. (2003a) investigated the metabolism of a series of 4-alkoxyanilines by bacterial NATs (S. typhimurium and M. smegmatis) and found that there was a marked increase in the rate of acetylation with increasing alkyl chain length (Hexyl > Bu > Et > Me). This result indicated that the lipophilicity of the substrate may be a contributory factor to the rate of acetylation, which is further confirmed by a strong correlation between the enzyme activities and the calculated partition coefficients (clogP) of the substrate molecules. The preference for acetylating more lipophilic substrates by the bacterial NATs is probably due to the lipophilic nature of three conserved phenylalanine residues in the active site (Brooke et al., 2003a). The dominant binding forces are con-

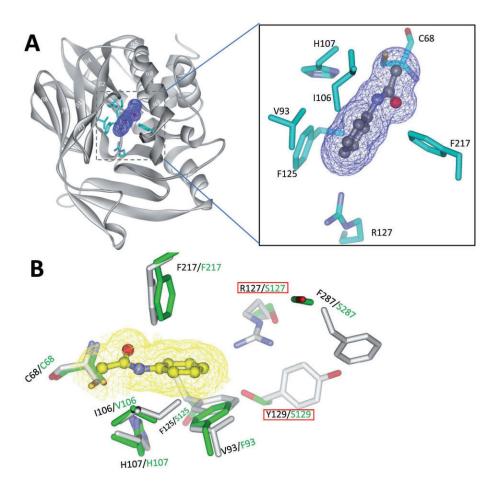


Figure 5

The substrate-binding site in human NAT. Panel A: A full view of the substrate-binding site (indicated as a blue surface) in human NAT (PDB code: 2PQT). Panel B: Comparison of the substrate-binding site residues of human NAT1 (in white; PDB code: 2PQT) with those of NAT2 (in green; PDB code: 2PFR). Residues at positions 127 and 129 play a critical role in determining the pocket size and the substrate specificity. Please see the text for details. The binding site is indicated as a yellow surface.

tributed by the lipophilic–lipophilic and π -stacking interactions between the phenylalanine residues in the enzyme and the acceptor substrate.

The crystal structure for M. smegmatis NAT (MSNAT)isoniazid (INH) complex has been used to explain the enzyme substrate selectivity towards hydrazines (HDZ) and arylamines (Sandy et al., 2005b). The arylhydrazines are the best substrates for MSNAT with relatively low K_m values of <5 μM for HDZ and 7.3 μM for INH, whereas the arylamines have higher K_m values - 586 µM for 5-aminosalicylic acid (5-AS), 1460 μM for ANS, 4500 μM for PAS, and 56 000 μM for PABA. This may be in part due to the formation of hydrogen bonds between the hydrazyl group and residues T109 and C70 (Figure 7B). In addition, HDZ has extra potential hydrogen bonding from the heterocyclic nitrogen to the backbone oxygen of G129 or F130 (Sandy et al., 2005b). The affinity of the arylamine series depends to some extent on the capacity to form interactions with T109 and F130. 5-AS is a good arylamine substrate because it can be well oriented to form hydrogen bonds with both the side chain of T109 and the main chain carbonyl of F130. PAS has a lower apparent affinity compared with 5-AS because it is difficult for T109 to form

hydrogen bonds with the substrate molecule, although hydrogen bonds to F130 can be readily achieved (Sandy $et\,al.$, 2005b). ANS has a lower K_m value than PAS despite offering hydrogen bonds to neither T109 nor F130. In this case, the unfavourable ligand binding in the absence of hydrogen bonding might be partly compensated by the generally lipophilic character of the methoxy group. Such groups are preferred in NAT substrates, presumably due to the formation of hydrophobic interactions in the active site. PABA has no additional chemical group on the aromatic ring facing the T109 and has an even higher K_m than that of PAS (Sandy $et\,al.$, 2005).

Structure–activity relationships for NAT inhibitors

There is considerable interest in developing small molecule inhibitors for NATs because (i) the inhibitors may be used for therapeutic purposes in breast cancer and tuberculosis; and (ii) the inhibitors can be used to explore the functions



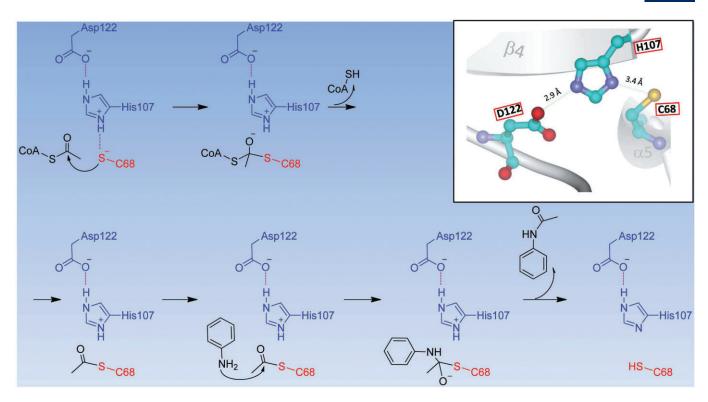


Figure 6

The proposed catalytic mechanism of NAT showing the importance of the catalytic triad Cys⁶⁸-His¹⁰⁷-Asp¹²² (human NAT numbering) in initiating the acetylation reaction. The insert depicts the location of the triad residues according to the crystal structure of human NAT2 (PDB code: 2PFR).

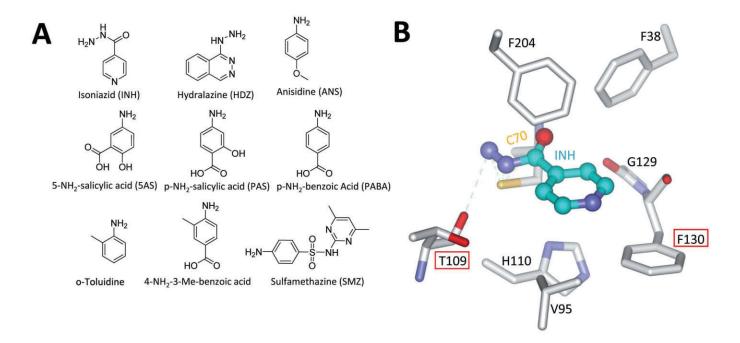


Figure 7

Substrate-binding site of NAT from M. smegmatis (MSNAT) and the structural determinants of its substrate preference. Panel A: Chemical structures of NAT substrates. Panel B: Diagram representation of MSNAT-isoniazid interactions (PDB code: 1W6F). Interactions with T109 and F130 are important in substrate binding.



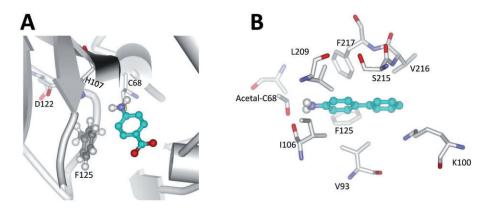


Figure 8

Predicted molecular interactions between p-aminobenzoic acid (PABA) and human NAT1 (Panel A; PDB code: 2DSS) and between 4-aminobiphenyl and human NAT1 (Panel B; PDB code: 2GWZ) based on the NMR model of human NAT1.

Figure 9

Chemical structures of three types of NAT inhibitors, rhodanine derivative (panel A), TZD-benzosultam derivative (panel B) and S-substituted 1,2,4-triazole (panel C). For TZD-benzosultam derivatives, the hydrophobicity of the N-substitution correlates with the binding affinity.

of NATs in various organisms (Westwood et al., 2011). Russell et al. (2009) synthesized and evaluated a series of rhodanine (2-sulfanylidene-1,3-thiazolidin-4-one) derivatives as selective inhibitors of human NAT1. The (Z)-5-(20hydroxybenzylidene)-2-thioxothiazolidin-4-one (Figure 9A) was one of the most potent inhibitors and this compound was docked into the crystal structure of human NAT1 to explore the inhibitor binding mode (Russell et al., 2009). The aryl substituent extends into a hydrophobic pocket surrounded by residues V93, F125, V216 and F287, providing a possible explanation for the preference for lipophilic substituents. Binding of the inhibitor is also stabilized by hydrogen bonds with residues R127 and T289. The additional hydrogen bonding between the 2'-hydroxyl of the inhibitor and the enzyme may provide an explanation for the relatively high potency of 2'-hydroxyl-substituted rhodanine over its 4'-hydroxyl-substituted counterpart (Russell et al., 2009).

Brooke *et al.* (2003a) identified the N-(2-naphthyl)-methyl substituted 1,1-dioxo-2,3-dihydrobenzo[1,2]-

thiazine-4-ylidene thiazolidine-2,4-dione [N-(2-naphthyl)methyl and 2,4-thiazolidinedione (TZD)-substituted benzosultam, Figure 9B] to be a weak inhibitor (IC₅₀ = 78 μ M) of MSNAT by high throughput screening of a proprietary small molecule library. Subsequent synthesis and screening of a series of TZD-sultam adducts revealed that the TZD moiety was essential for activity as well as the substitution on the sultam nitrogen (Brooke et al., 2003b). The most effective NAT inhibitors in this series were those incorporating hydrophobic substituents at the sultam nitrogen. Docking of the TZD-sultam adducts with the crystal structure of MSNAT revealed the inhibition mechanism and the detailed molecular interactions between the inhibitor and the enzyme. The sultam moiety was located adjacent to the catalytic residue cysteine, providing a rationale for the observed competitive inhibition. The substituents at the sultam nitrogen occupied a hydrophobic groove in the protein, with a tryptophan residue, W97, appearing to hold the compound in place (Brooke et al., 2003b). The imidazole moiety of H203 participated in a π -stacking interaction with the TZD moiety of the



Table 2 Mutation sites of human NAT alleles and their location in the NAT proteins as well as the functional effects of the mutations

NAT	Mutation site	<i>NAT</i> Allele	Location in the protein	Functional effect	Reference
NAT1	R117	NAT1*5	On the surfaces	Mutants may be subject to increased ubiquitinylation,	Wu <i>et al.,</i> 2007
	V149	NAT1*11A	of the protein	leading to reduced protein level and reduction in the	Hein, 2002 Liu <i>et al.</i> , 2006 Walraven <i>et al.</i> , 2008a
R166 M205 S214 E261 R64 E167 R187		NAT1*11B		enzymic activity. Neither the V149I nor the S214A residue changes alter the structural stability of NAT1. No functional changes occur with M205V and E261K mutations.	
		NAT1*30			
	R166	NAT1*5			
		NAT1*21			
	S214	NAT1*11A			
		NAT1*11B			
		NAT1*11C			
	E261	NAT1*24			
		NAT1*17	On the α4-α5	R64 forms H-bonds with the neighbouring residues E38	Wu et al., 2007
		NAT1*19B	loop	and N41. The stability of the enzyme is compromised in the absence of these interactions.	Walraven et al., 2008
	E167	NAT1*5	At the beginning of β10	E167 forms H-bonds with the neighbouring residues K185 and D251. The mutant may affect protein stability.	Wu et al., 2007
	R187	NAT1*14A NAT1*14B	In the 17-residue insertion	R187 forms an H-bond with E182. Substitution of R187 most likely decreases protein stability and lowers protein levels. The mutant may also alter the active site topology.	Wu <i>et al.</i> , 2007 Hughes <i>et al.</i> , 1998
	D251	NAT1*22	On the strand β15	D251 forms H-bonds with the neighbouring residues R242 and N245. The mutant may break these interactions and result in destabilization of the protein.	Wu <i>et al.,</i> 2007 Hein, 2002 Lin <i>et al.,</i> 1998
	1263	NAT1*25	In the α 11	No change in protein level or catalytic activity for the 1263V mutant because the hydrophobic interactions of the residue with others are preserved without introducing steric clashes.	Walraven <i>et al.,</i> 2008
NAT2	l114	NAT2*5 NAT2*14C/F	On the surfaces of the protein	Mutants may be subject to increased ubiquitinylation, leading to reduced protein level and reduction in the enzymic activity.	Wu et al., 2007 Hein, 2002 Liu et al., 2006
	E167	NAT2*10			
	R197	NAT2*5E/J			
		NAT2*6			
		NAT2*14D			
	K268	NAT2*5			
	NZOO	NAT2*6C/F			
		NAT2*12			
		NAT2*14C/E-G/I			
	K282	NAT2*18			
	G286	NAT2*61/J			
	3200	NAT2*7			
	R64	NAT2*7D	On the α4-α5	R64 forms H-hands with the neighbouring residues 529	Wu <i>et al.,</i> 2007
	NO4	loon		R64 forms H-bonds with the neighbouring residues E38 and N41. The stability of the enzyme is compromised in the absence of these interactions.	Wu <i>et al., 2007</i> Walraven <i>et al.,</i> 2008
		NAT2*14 NAT2*19	1		
	D122	NAT2*12D	On the β5-α6 loop	D122 is a member of the catalytic triad. Mutations of D122 would adversely affect the activity of the enzyme.	Wu <i>et al.,</i> 2007 Walraven <i>et al.,</i> 2008
	L137	NAT2*51	On the β6-β7 loop	L137 makes contacts with residues L194 and W159 through hydrophobic interactions. The mutant may result in a change in secondary structure that could trigger degradation mechanisms.	Wu <i>et al.,</i> 2007 Walraven <i>et al.,</i> 2008
	Q145	NAT2*17	On the β7-β8 loop	Q145 forms H-bonds with the neighbouring residues W132 and Q133. The mutant shows lower enzymic activity that may be due to reduced expression levels.	Hein, 2002 Wu <i>et al.</i> , 2007

compounds, which is consistent with the observed inactivity of the precursor ketone and ketal derivatives (Brooke *et al.*, 2003b).

A more recent study characterized the inhibitory potency of a series of 1,2,4-triazoles (with long chain aliphatic or planar aromatic substituents on sulfur; Figure 9C) against M. tuberculosis NAT (Westwood et al., 2010). Interestingly, the potency of NAT inhibition increased with extended chain length up to approximately 10 Å, with the S-4-phenylbenzyl and S-octyl derivatives being the most potent in the series. The S-decyl derivative has an approximate chain length of 13 Å and was a less potent inhibitor of NAT than the shorter chain S-alkyl derivatives. The IC₅₀ value against NAT of the S-decyl derivative was 13.1 μM, compared with an IC₅₀ value of 3.8 μM for the S-octyl-1,2,4-triazole. This trend in acetylation difference was explored by molecular docking of the S-substituted 1,2,4-triazoles with NAT from M. tuberculosis. All except the S-decyl derivative were predicted to bind in essentially identical orientations. The triazole ring formed π-stacking interactions with F130 and formed hydrogenbonds with the backbone carbonyls of F130 and G131. The 4'-methylphenyl group in all molecules (except the S-decyl derivative) pointed towards the active site cysteine residue. The correlation observed between the chain length and NAT inhibitory potency was closely matched by the docking scores for the same series of compounds, indicating that the crystal structures for NATs are powerful tools for structural design of inhibitors and for understanding of the affinity differences of the ligands.

NAT Polymorphisms

Acetylation was the first enzymic polymorphism to be investigated and was based on the observation that the antitubercular drug isoniazid (INH) caused a different level of neural toxicity across different populations (Grant et al., 1997; Walker et al., 2009). NAT polymorphisms have been linked to three types of acetylator phenotypes, namely, fast, intermediate and slow acetylators (Hein et al., 2000; Hein, 2006). It has been suggested that slow acetylators are most at risk from INH hepatotoxicity because the toxic metabolite acetylhydrazine is retained much longer in the liver, compared with fast acetylators (Ellard et al., 1978; Metushi et al., 2011). In NAT1, there are 27 variant alleles that are associated with 11 amino acid alterations, whereas in NAT2, there are 65 variant alleles associated with alterations of 22 amino acids. A complete list of the NAT alleles and their associated phenotypes can be obtained online (http://louisville.edu/ medschool/pharmacology). Understanding of why and how the polymorphisms alter the activity of human NATs has been significantly advanced by the availability of the crystal structures for the enzymes (Wu et al., 2007). Based on the location and/or functional role of the original amino acids in the protein, the functional effects of NAT polymorphisms are well elucidated (Wu et al., 2007). Table 2 summarizes the correlations of the structural data and functional implications for the NAT mutations. For more detailed discussion, one is encouraged to read the reviews by Walraven et al. (2008a,b).

Conclusion

NAT plays an important role in the biotransformation of many aromatic and heterocyclic amine drugs. In addition, it has been linked to cancer risk because of its roles in the metabolic activation of carcinogens and in cell growth and survival. This review has described the crystal structures for NATs and analysed the structural similarities and differences between prokaryotic and mammalian NATs. Furthermore, breakthroughs in understanding of the catalytic mechanism, substrate/inhibitor binding and polymorphisms were discussed from a structural perspective. Although the crystal structures represent a powerful tool for predicting substrate selectivity, they have also been very useful in rational design of selective inhibitors of NATs with great potential in cancer treatment

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Conflict of interest

The authors report no conflict of interest.

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